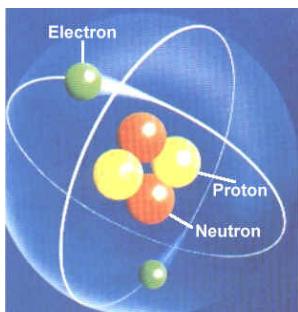


# Spectroscopy on Atoms and selection rules

## I - What is an atom? Modeling ?



- N-electrons bound to a nucleus ( $Z$ )

$$H_{n-e} = \sum_{i=1 \dots N} -\frac{\hbar^2}{2m} \nabla_{\vec{r}_i}^2 - \frac{Ze^2}{4\pi\epsilon_0 r_i}$$

- N-electrons in interactions

$$H_{e-e} = \sum_{i < j = 1 \dots N} \frac{e^2}{4\pi\epsilon_0 r_{ij}}$$

- N-electron atom wave function  $\Psi(q_1 \dots q_N)$  with  $q_i = \{\vec{r}_i, \sigma_i\}$  satisfies

$$\begin{aligned} H|\Psi(q_1 \dots q_N)\rangle &= E|\Psi(q_1 \dots q_N)\rangle \\ \Leftrightarrow [H_{e-n} + H_{e-e}]|\Psi(q_1 \dots q_N)\rangle &= E|\Psi(q_1 \dots q_N)\rangle \end{aligned}$$



# Step 1: Independent-particle model – no spin

- Lets introduce the spherically symmetric potential  $V(r)$

$$V(r) = -\frac{Ze^2}{4\pi\epsilon_0 r} + S(r) \quad \text{with} \quad \begin{cases} V(r) \rightarrow -\frac{Ze^2}{4\pi\epsilon_0 r} & r \rightarrow 0 \\ V(r) \rightarrow -\frac{(Z-N+1)e^2}{4\pi\epsilon_0 r} & r \rightarrow \infty \end{cases}$$

Screening of the nucleus

- Re-writing the total Hamiltonian

$$H = H_{e-n} + H_{e-e} + \sum_{i=1 \dots N} V(r_i) - \sum_{i=1 \dots N} V(r_i)$$

$\Rightarrow H = H_{central} + H_{noncentral}$

$$H_{central} = \sum_{i=1 \dots N} -\frac{\hbar^2}{2m} \nabla_{\vec{r}_i}^2 + V(r_i)$$

$$H_{noncentral} = H_{e-e} - \sum_{i=1 \dots N} S(r_i)$$

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# Step 1: Independent-particle model – no spin

- The Schrodinger equation for the central field is separable in  $N$  equations

$$H_{central} |\Psi_{central}(\vec{r}_1 \dots \vec{r}_N)\rangle = E_{central} |\Psi_{central}(\vec{r}_1 \dots \vec{r}_N)\rangle$$

since all particles are independent

$$|\Psi_{central}(\vec{r}_1 \dots \vec{r}_N)\rangle = \underbrace{|u_{\alpha_1}(\vec{r}_1)\rangle \otimes |u_{\beta_2}(\vec{r}_2)\rangle \dots \otimes |u_{\gamma_N}(\vec{r}_N)\rangle}_{\text{Set of good quantum numbers}}$$

- Schrodinger equation in a central field for the electron  $i$

$$\left( -\frac{\hbar^2}{2m} \nabla_{\vec{r}_i}^2 - V(r_i) \right) |u_{n_i l_i m_i}(\vec{r}_i)\rangle = E_{n_i l_i} |u_{n_i l_i m_i}(\vec{r}_i)\rangle$$

with the “natural” solution

$$|u_{n_i l_i m_i}(\vec{r}_i)\rangle = |R_{n_i l_i}(r_i)\rangle \otimes |Y_{l_i m_i}(\theta_i, \varphi_i)\rangle$$

*The radial part is not the solution of the hydrogen.  
The different  $l$ -states are not degenerate*

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# Step 1: Independent-particle model

- **Spin and the Pauli exclusion principle**

Electrons are fermions of spin  $1/2$ , so that the wave function  $\Psi(q_1 \dots q_N)$  must be anti-symmetric in the spatial and spin coordinates  $q_i$  of the electrons

- **Building up a wave function for N electrons – Slater determinants**

$$|\Psi_{central}(q_1 \dots q_N)\rangle = \frac{1}{\sqrt{N!}} \begin{vmatrix} u_\alpha(q_1) & u_\beta(q_1) & \dots & u_\nu(q_1) \\ u_\alpha(q_2) & u_\beta(q_2) & \dots & u_\nu(q_2) \\ \dots & \dots & \dots & \dots \\ u_\alpha(q_N) & u_\beta(q_N) & \dots & u_\nu(q_N) \end{vmatrix}$$

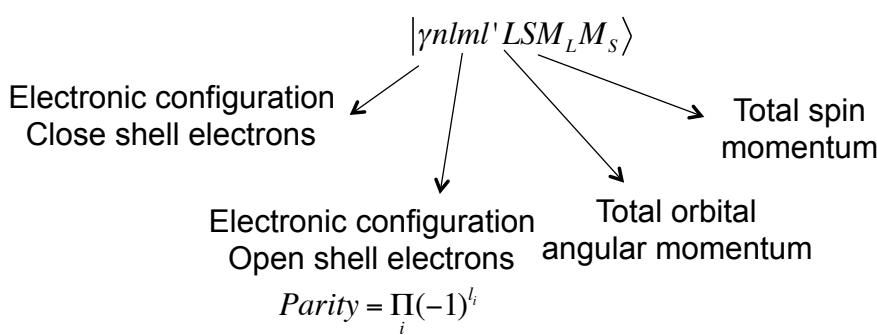
with  $|u_{n_l l_i m_l m_{s_l}}(q_i)\rangle = |R_{n_l l_i}(r_i)\rangle \otimes |Y_{l_i m_{l_i}}(\theta_i, \varphi_i)\rangle \otimes |\chi_{1/2, m_{s_l}}\rangle$



# Step 1: Independent-particle model

- **Basis set to describe an electronic state – LS representation**

$$\left\{ \begin{array}{l} \vec{L} = \sum_{i=1 \dots N} \vec{l}_i \\ \vec{S} = \sum_{i=1 \dots N} \vec{s}_i \end{array} \right.$$



- **Russel-Saunders notation**

$$|\gamma n l m l' [^{2s+1}L_J]\rangle$$



## Step 2: Beyond the independent-particle model

- Apply a perturbation  $W$  on the operator  $H_0$   $H_0|\varphi_n^0\rangle = E_n^0 |\varphi_n^0\rangle$

First order

$$\left[ \begin{array}{l} |\varphi_n^1\rangle = |\varphi_n^0\rangle + \sum_{p \neq n} \frac{|\langle \varphi_p^0 | W | \varphi_n^0 \rangle|^2}{E_n^0 - E_p^0} |\varphi_p^0\rangle \\ E_n^1 = E_n^0 + \langle \varphi_n | W | \varphi_n \rangle \end{array} \right]$$

Second order is required if  $\Delta E_{p,n}^0 = E_p^0 - E_n^0 \approx \langle \varphi_p | W | \varphi_n \rangle$

- To be considered in the Hamiltonian

$H_{\text{noncentral}} = H_{e-e} - \sum_{i=1 \dots N} S(r_i)$	$H_{SO} = \frac{1}{2(mc)^2} \sum_{i=1 \dots N} \frac{1}{r_i} \frac{dV(r_i)}{dr_i} \vec{l}_i \cdot \vec{s}_i$	
 Electronic correlation ( $\propto Z^2$ )	 Spin orbit ( $\propto Z^4$ )	 <i>and more terms...</i>

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## Step 2: Beyond the independent-particle model

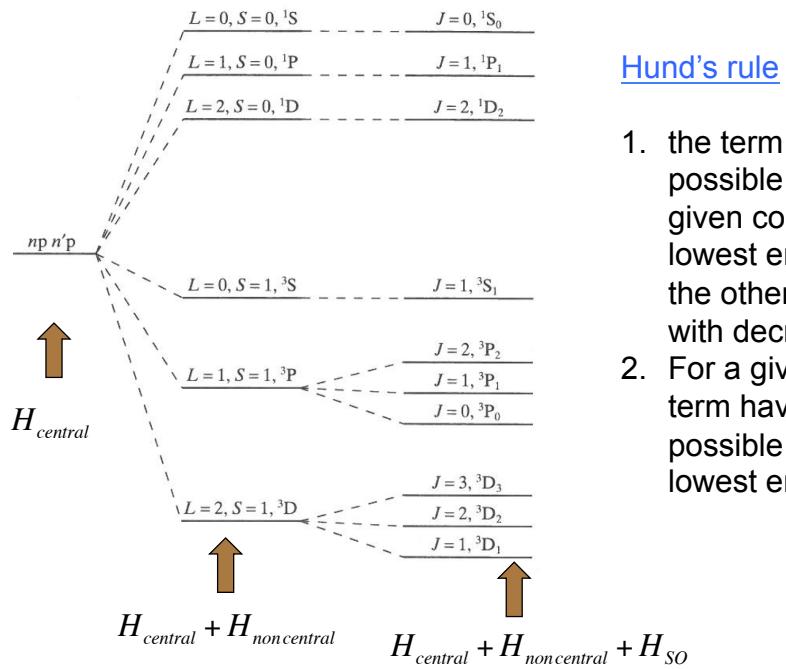
	$H_{\text{noncentral}} \gg H_{SO}$	$H_{SO} \gg H_{\text{noncentral}}$
First perturbation	$H = H_{\text{central}} + H_{\text{noncentral}}$ $[H, \vec{L}] = [H, \vec{S}] = 0$	$H = H_{\text{central}} + H_{SO}$ $[H, \vec{J}] = 0$
	 <b>LS - coupling</b>	 <b>JJ - coupling</b>
Second perturbation	$H + H_{SO}$ $[H, \vec{J}] = 0$	$H + H_{\text{noncentral}}$ $[H, \vec{J}] = 0$
Dipole Selection rules	$\Delta J = 0, \pm 1 \text{ with } J_i = J_f = 0 \text{ forbidden}$ $\Delta S = 0$ $\Delta L = 0, \pm 1 \text{ with } L_i = L_f = 0 \text{ forbidden}$	$\Delta j = 0, \pm 1$ $\text{with } j_i = j_f = 0 \text{ forbidden}$

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## Step 2: Beyond the independent-particle model

### Fine structure with LS-coupling



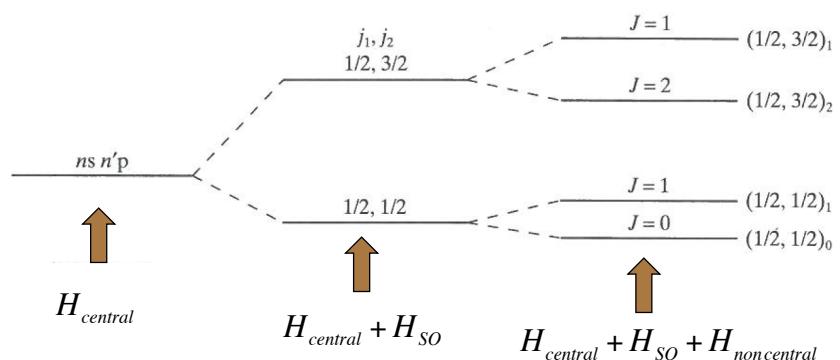
### Hund's rule

1. the term with the largest possible value of S for a given configuration has the lowest energy; the energy of the other terms increases with decreasing S
2. For a given value of S, the term having the maximum possible value of L has the lowest energy



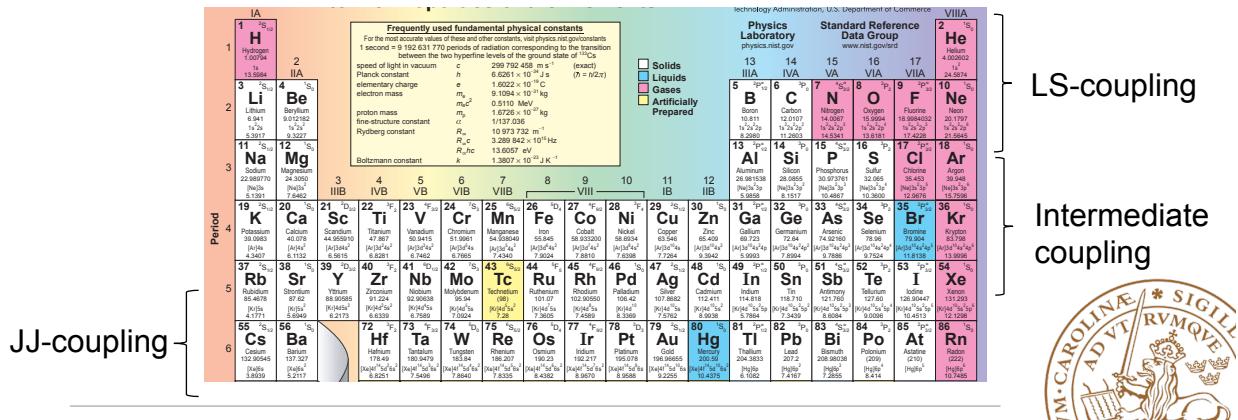
## Step 2: Beyond the independent-particle model

### Fine structure with JJ-coupling



# Summary

- Define a basis to describe the **electronic structure**
  - simple model : independent-particle model
- Use this basis to describe the **fine structure**
  - more sophisticated model including electron correlations, relativistic term
  - coupling scheme informs on the interactions



## II - Experimental manifestation

V. Schmidt, Rep. Prog. Phys. **55**, 1483 (1992)

### Absorption Cross sections

$$\sigma_a = \sum_{\text{all } n_f} \{\sigma^+ + \sigma^{+*} + \sigma^{++} + \dots\}$$

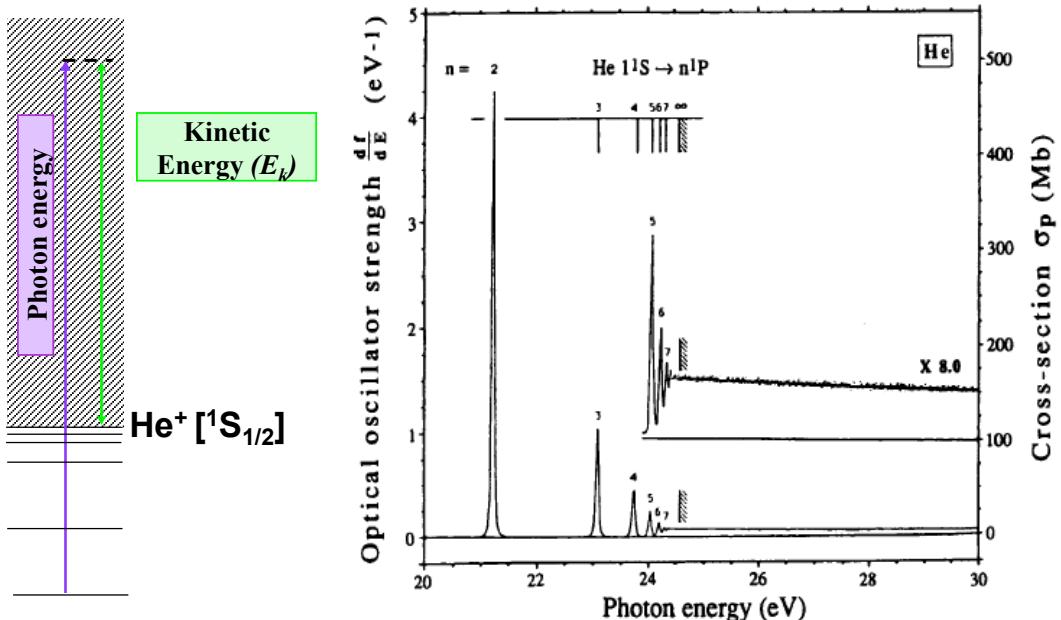
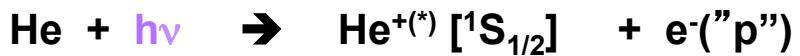
**Main transitions:** single ionization cross section ( $\sigma^+$ ) for different shells

**Satellite transitions:** many electron processes – mainly two electrons double excitation ( $\sigma^{**}$ ), ionization and excitation ( $\sigma^{+*}$ -discrete satellites) double ionization ( $\sigma^{++}$ - continuous satellites)

$$\rightarrow \sigma_a = \sigma_{\text{ionization}} + \sigma_{\text{fluorescence}} (+\sigma_{\text{dissociation}})$$



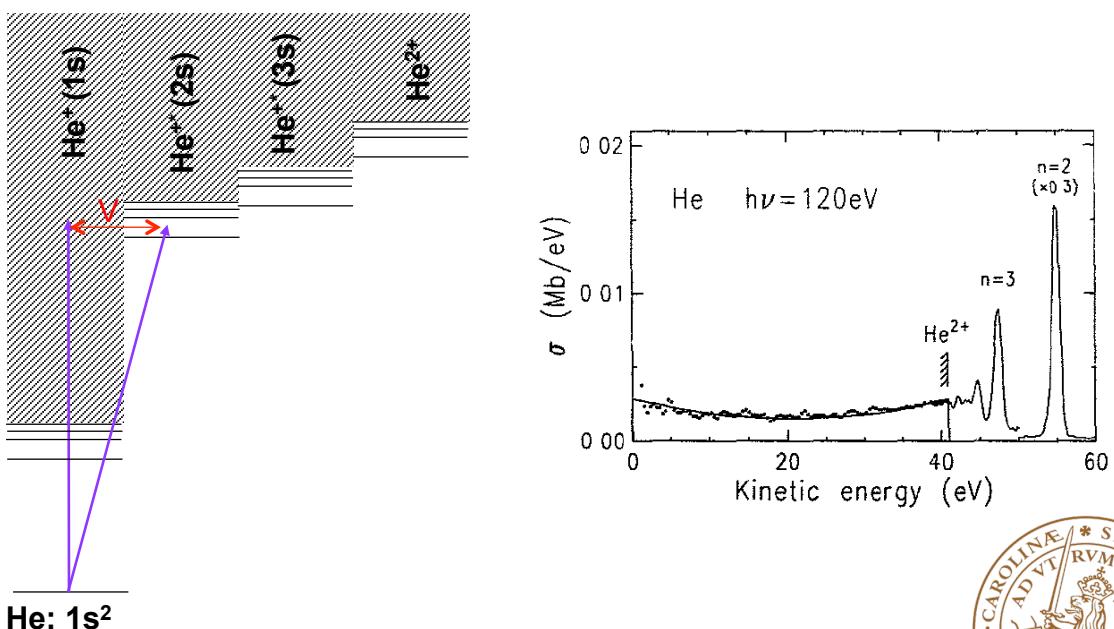
# Photo-absorption / Photo-ionization



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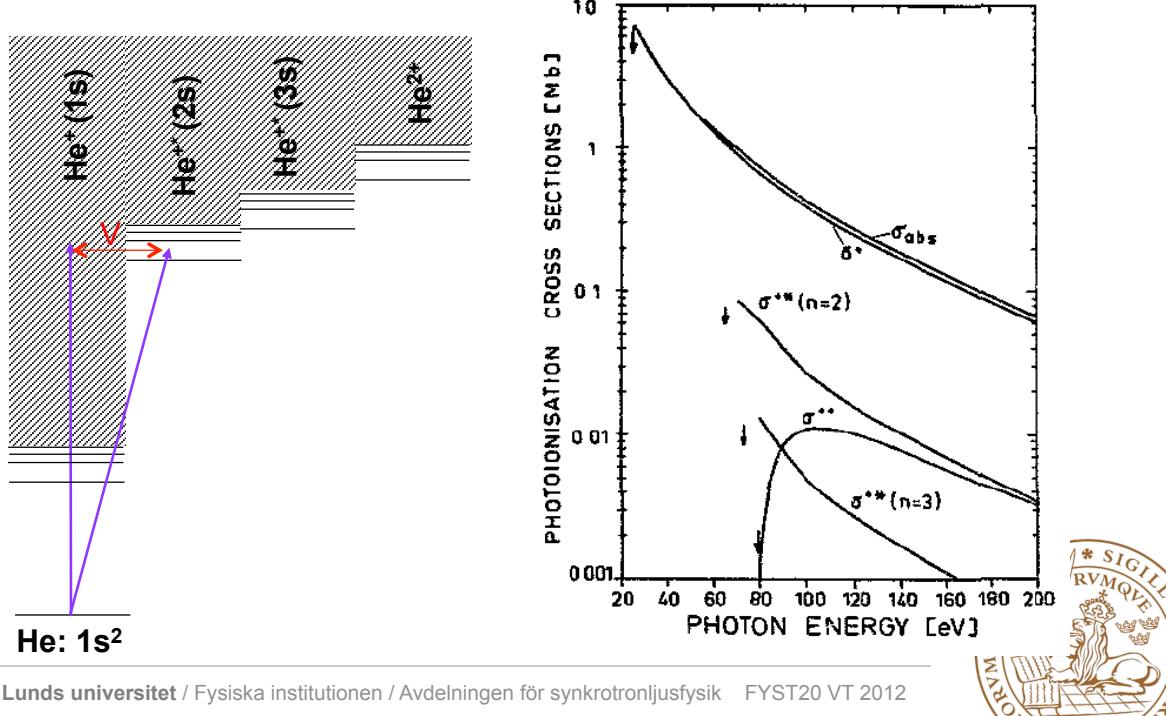
# Total cross section / partial cross- section



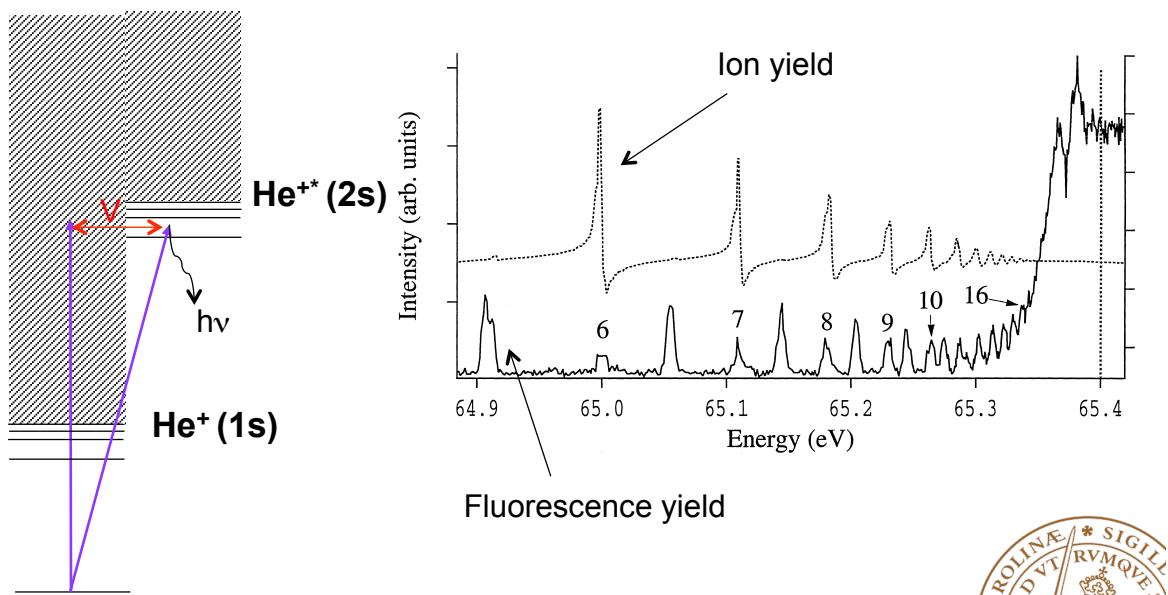
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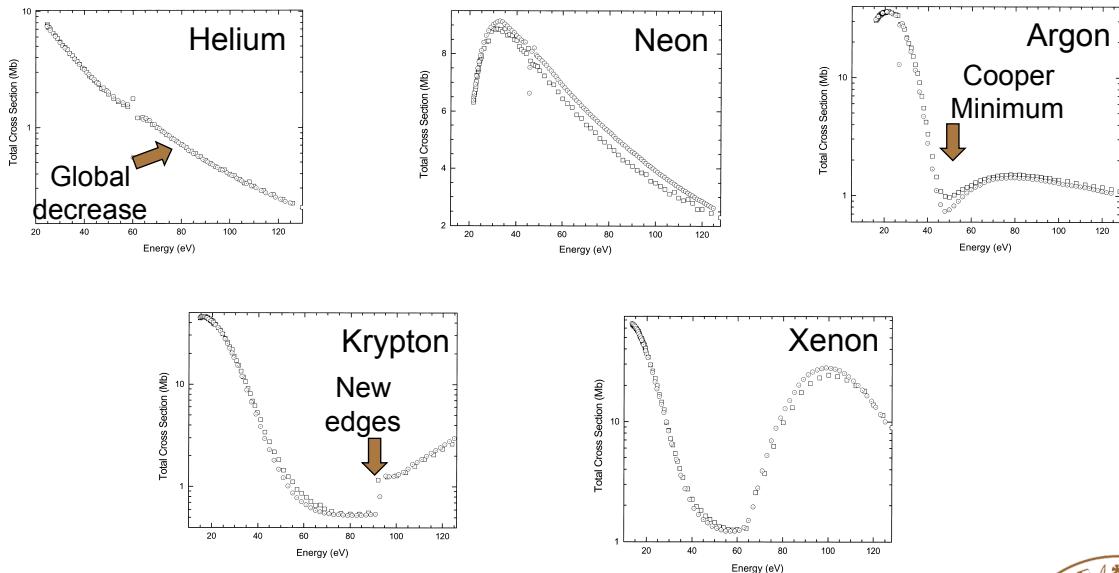
# Total cross section / partial cross-section



## Photo-absorption/ ionization & Fluorescence



# Evolution of the cross section



Samson *et al*, J. Elec. Spec. Rel. Phenom, 123 ,265 (2002)

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## Effect of the radial overlap

Many electron matrix element  
in terms of one electron matrix element

$$D_\gamma = (2j_1 + 1)^{1/2} i^{-l_\gamma} (-)^{l_\gamma - 1/2} e^{i(\sigma_\gamma + \delta_\gamma)} R_{\gamma, nl, j_1} (2j + 1)^{1/2} \begin{pmatrix} j & 1 & j_1 \\ \frac{1}{2} & 0 & -\frac{1}{2} \end{pmatrix}.$$



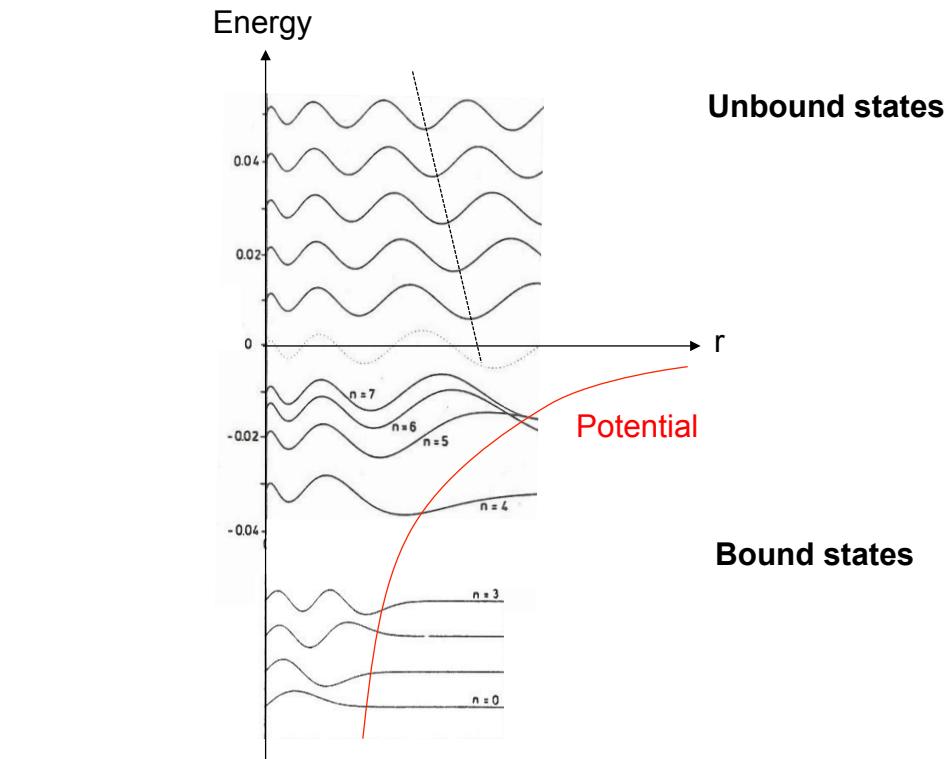
Radial overlap

$$R_{\gamma, nl, j_1} = \int_0^\infty P_\gamma(r) r P_{nl, j_1}(r) dr$$



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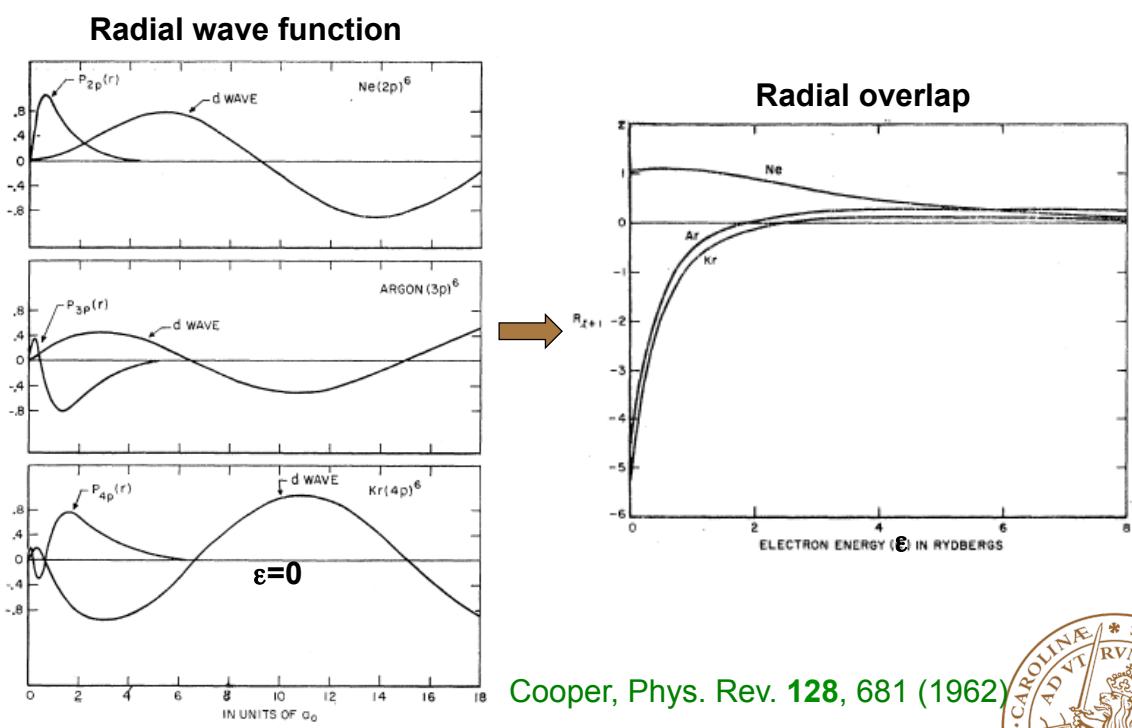
## a) Global decrease



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## b) Cooper minima

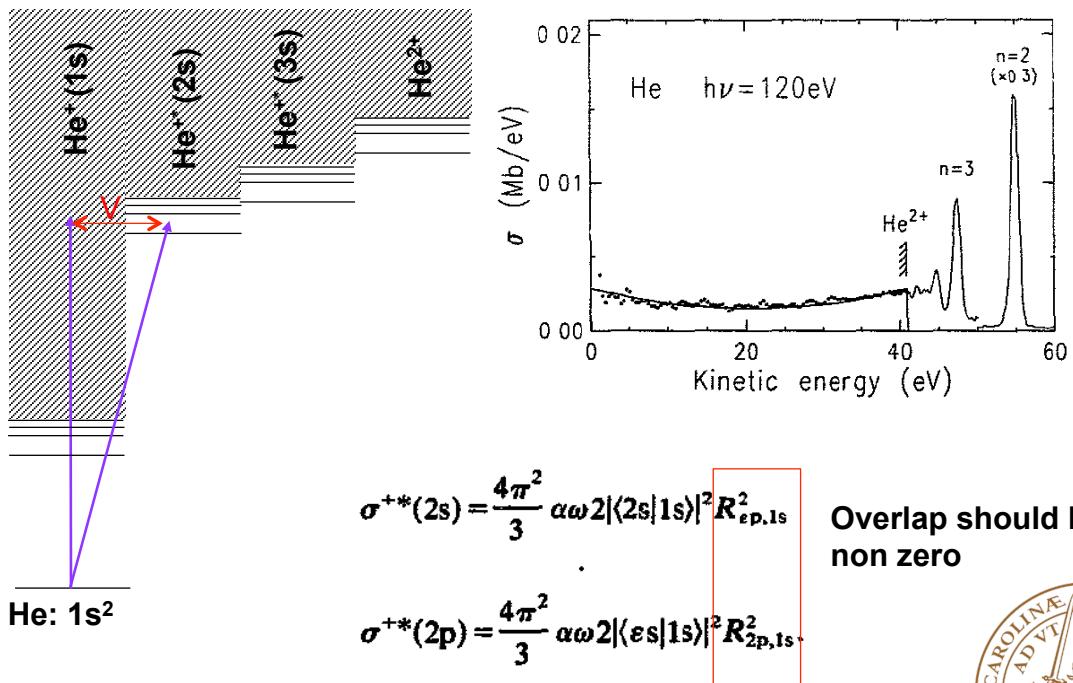


Cooper, Phys. Rev. **128**, 681 (1962)

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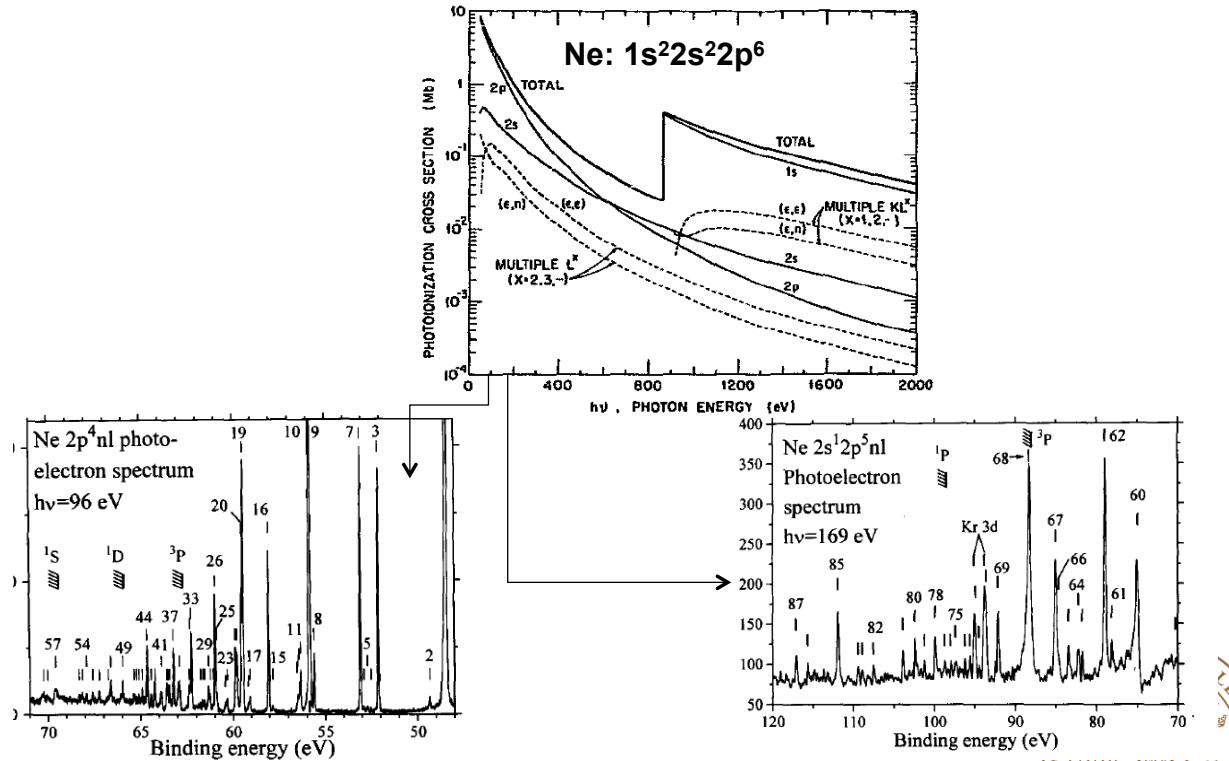
# Satellites – Shake up



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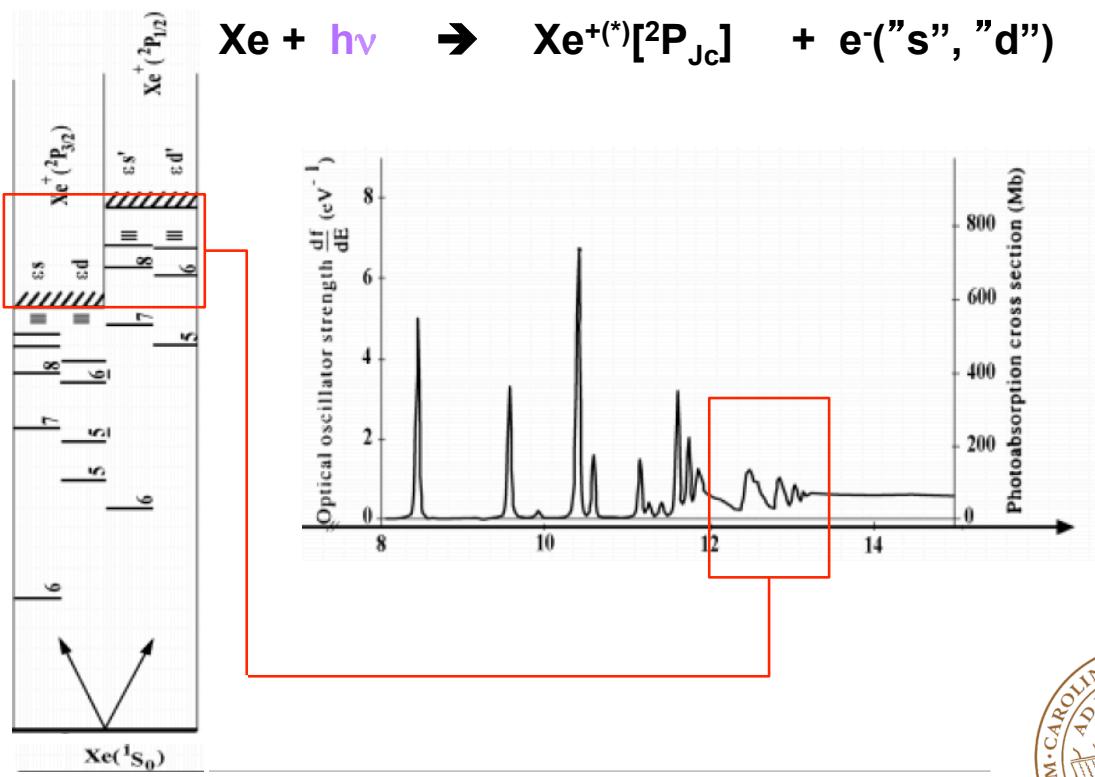
# Total cross section / partial cross- section



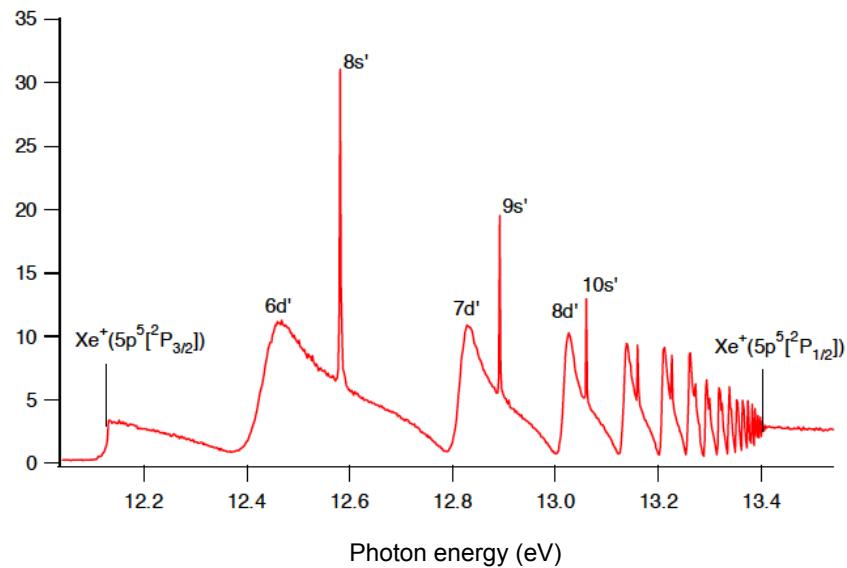
... and some more effects..?..



## Photo-absorption / Ionization – two cores



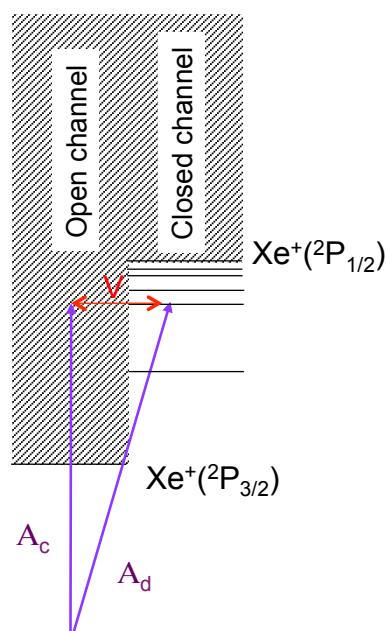
# Auto-ionization – two cores



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# Auto-ionization – parametrization



Parameterization of the resonance

$$\sigma(E) = \sigma_a \frac{(q + \varepsilon)^2}{1 + \varepsilon^2}$$

$\sigma_a$ : is the cross section  $\propto |A_c|^2$

$q$ : is the profile index  $\propto A_d / (V^* A_c)$

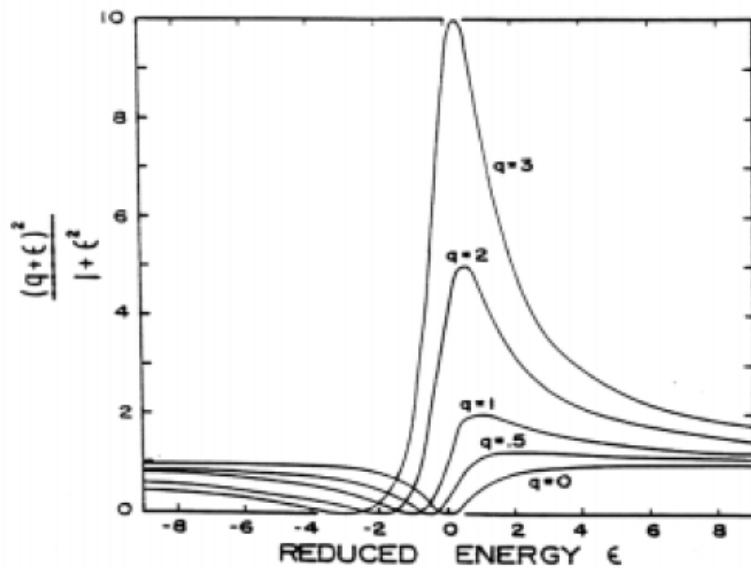
$\varepsilon$ : is the reduced energy  $(E - E_r) / (\Gamma / 2)$

$\Gamma$ : is resonance width  $2\pi|V_E|^2$

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# Auto-ionization - parametrization



U. Fano, Phys. Rev. **124**, 1866 (1961)



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## Tips: Rydberg series assignment

### Generalized Rydberg formulae

$$E_n = IP - \frac{R}{(n - \mu)^2} = IP - \frac{R}{n^{*2}}$$

IP: Ionization potential

R: Rydberg constant ( $109736,86 \text{ cm}^{-1}$ )

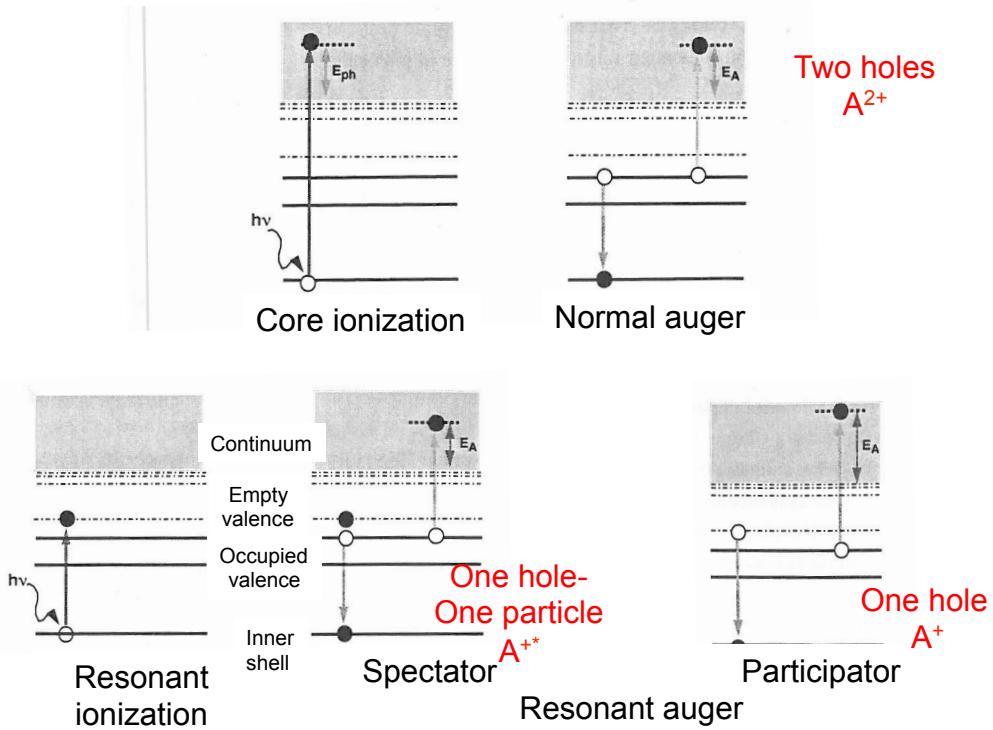
n: principal quantum number,  $n^*$  effective principal quantum number

$\mu$ : quantum defect



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# Auger



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